

## 2 Decimals, standard form, and significant figures

### 2.1 Decimal numbers

A decimal number has a decimal point. Each figure *before* the point is a whole number, and the figures *after* the point represent fractions.

The number of decimal places is the number of figures *after* the decimal point. For example, the number 47.38 has 2 decimal places, and 47.380 is the same number to 3 decimal places.

In science, you must write your answer to a sensible number of decimal places.

#### Practice questions

- 1 New antibiotics are being tested. A student calculates the area of clear zones in Petri dishes in which the antibiotics have been used. List these in order from smallest to largest.

0.0214 cm<sup>2</sup>      0.03 cm<sup>2</sup>      0.0218 cm<sup>2</sup>      0.034 cm<sup>2</sup>

- 2 A student measures the heights of a number of different plants. List these in order from smallest to largest.

22.003 cm      22.25 cm      12.901 cm      12.03 cm      22 cm

### 2.2 Standard form

Sometimes biologists need to work with numbers that are very small, such as dimensions of organelles, or very large, such as populations of bacteria. In such cases, the use of scientific notation or standard form is very useful, because it allows the numbers to be written easily.

Standard form is expressing numbers in powers of ten, for example,  $1.5 \times 10^7$  microorganisms.

Look at this worked example. The number of cells in the human body is approximately 37 200 000 000 000. To write this in standard form, follow these steps:

**Step 1:** Write down the smallest number between 1 and 10 that can be derived from the number to be converted. In this case it would be 3.72

**Step 2:** Write the number of times the decimal place will have to shift to expand this to the original number as powers of ten. On paper this can be done by hopping the decimal over each number like this:

6.3900000000

until the end of the number is reached.

In this example that requires 13 shifts, so the standard form should be written as  $3.72 \times 10^{13}$ .

For very small numbers the same rules apply, except that the decimal point has to hop backwards. For example, 0.000 000 45 would be written as  $4.5 \times 10^{-7}$ .

#### Practice questions

- 3 Change the following values to standard form.

a 3060 kJ

b 140 000 kg

c 0.000 18 m

d 0.000 004 m

4 Give the following numbers in standard form.

a 100

b 10 000

c 0.01

d 21 000 000

5 Give the following as decimals.

a  $10^6$

b  $4.7 \times 10^9$

c  $1.2 \times 10^{12}$

d  $7.96 \times 10^{-4}$

### 2.3 Significant figures

When you use a calculator to work out a numerical answer, you know that this often results in a large number of decimal places and, in most cases, the final few digits are 'not significant'. It is important to record your data and your answers to calculations to a reasonable number of significant figures. Too many and your answer is claiming an accuracy that it does not have, too few and you are not showing the precision and care required in scientific analysis.

Numbers to 3 significant figures (3 s.f.):

7.88   25.4   741

Bigger and smaller numbers with 3 significant figures:

0.000 147   0.0147   0.245   39 400   96 200 000 (notice that the zeros before the figures and after the figures are *not* significant – they just show you how large the number is by the position of the decimal point).

Numbers to 3 significant figures where the zeros *are* significant:

207   4050   1.01 (any zeros between the other significant figures *are* significant).

Standard form numbers with 3 significant figures:

$9.42 \times 10^{-5}$     $1.56 \times 10^8$

If the value you wanted to write to 3 s.f. was 590, then to show the zero was significant you would have to write:

590 (to 3 s.f.) or  $5.90 \times 10^2$

**Remember:** For calculations, use the same number of figures as the data in the question with the lowest number of significant figures. It is not possible for the answer to be more accurate than the data in the question.

### Practice question

6 Write the following numbers to **i** 2 s.f. and **ii** 3 s.f.

a 7644 g

b 27.54 m

c 4.3333 g

d  $5.995 \times 10^2 \text{ cm}^3$

7 The average mass of oxygen produced by an oak tree is 11800 g per year.

Give this mass in standard form and quote your answer to 2 significant figures.